

Welcome

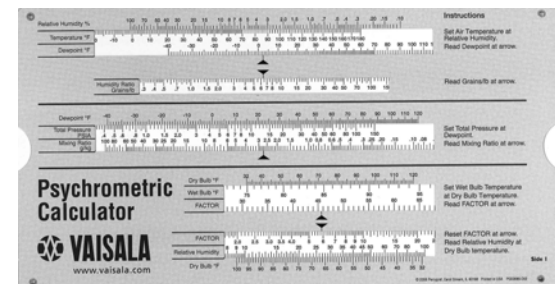
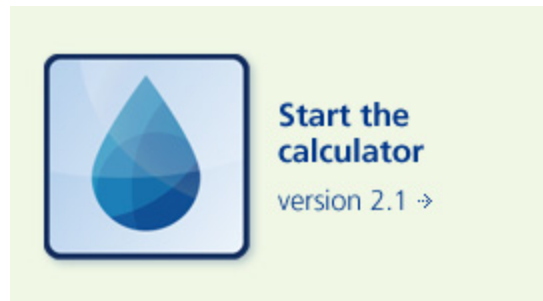
Vaisala Humidity 101 – Humidity Theory, Terms & Definitions
Part 1 of 3

Panelists: Bruce McDuffee, National Humidity Instructor
 Yumi Shakya, Application Engineer

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Agenda

1. Understand the basic theory of humidity
2. Dalton's Law
3. Vapor pressures
4. Relative humidity
5. At the end – give out some links to some great humidity resources



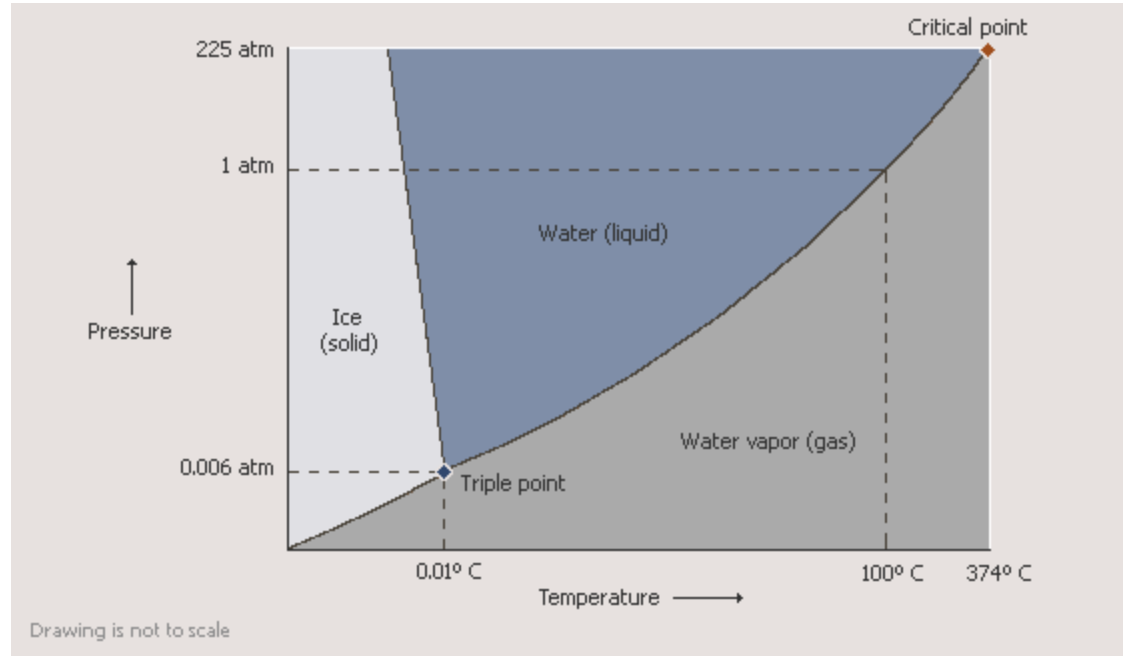
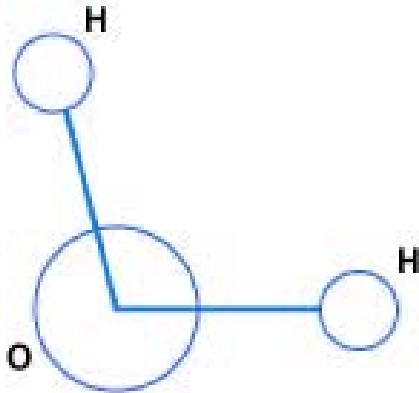
Humidity 101

\ webinar series

Humidity Theory, Terms & Definitions (Part 1 of 3)

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Water Vapor Theory - H₂O



- Exists in the three phases
- Which phase depends on the amount of thermal energy that is present

American Meteorological Society Glossary Hu-mid-i-ty

Humidity

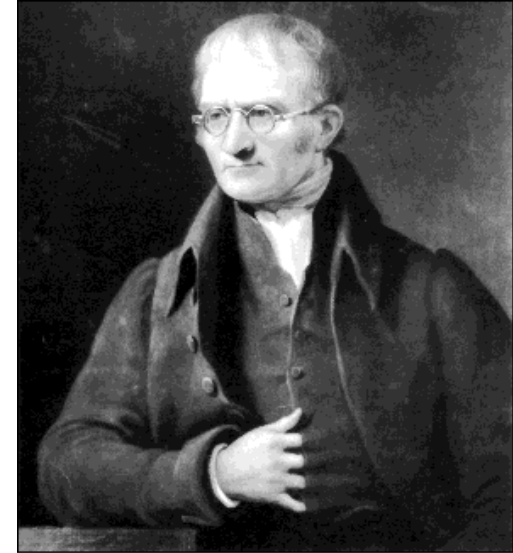
1. Generally, some measure of the water vapor content of air.



Dalton's Law

“The total pressure of a gas is equal to the sum of the different gases’ partial pressures”

$$P_t = P_1 + P_2 + \dots P_n$$



John Dalton

air around us

$$P_t = P_{N_2} + P_{O_2} + P_w + P_{misc.}$$

Practical Example of Dalton's Law

Nitrogen.....	77%
Oxygen	21%
Water vapor	1%
Other gasses	1%

$$1000 \text{ mbar} = 770\text{mbar} + 210\text{mbar} + 10\text{mbar} + 10\text{mbar}$$

In Denver

$$P_t = 840 \text{ mbar} \quad \begin{matrix} 840 \times 77\% & 840 \times 21\% & 840 \times 1\% & 840 \times 1\% \\ \underline{647} \text{ N}_2 + \underline{177} \text{ O}_2 + \underline{8} \text{ P}_w + \underline{8} \text{ Other} \end{matrix}$$

$$P_t = P_w + P_{\text{dry}}$$

Definitions

Psychrometry

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Partial Pressure of Water Vapor (psi,mbar,hPa,inhg...)

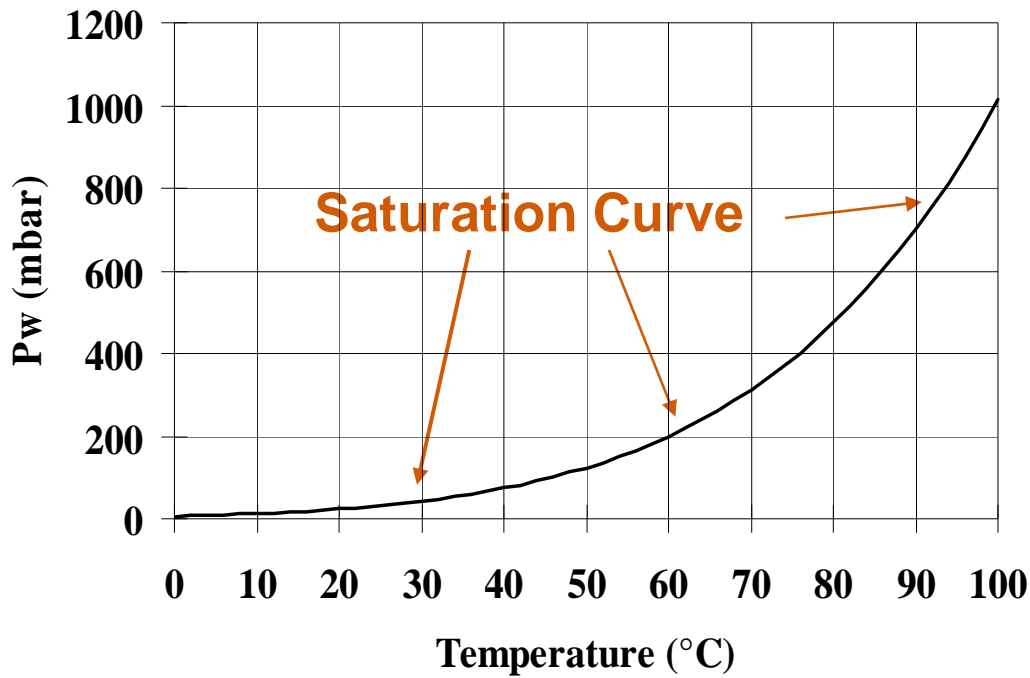
P_w

“The key parameter that affects all other humidity parameters”

- Note: The only two properties that can affect a change in P_w
- adding or removing water vapor
 - changes in system pressure

Saturation Vapor Pressure (psi,mbar,hPa,in hg...)

P_{ws}



On the saturation curve

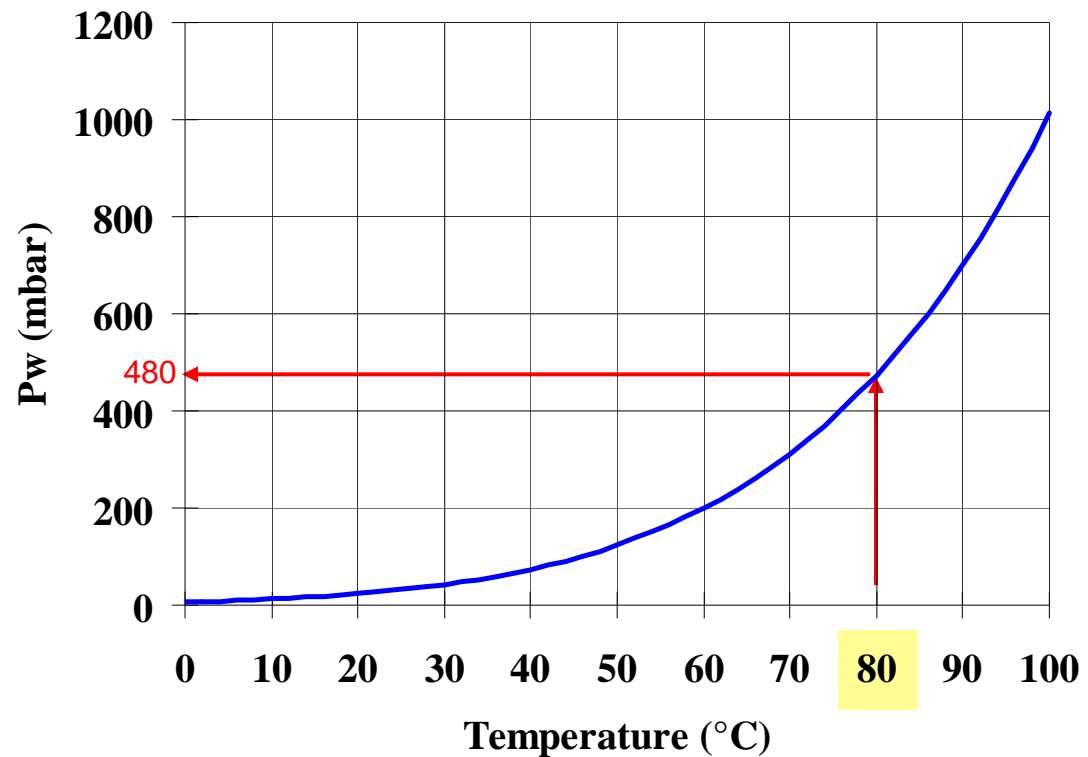
- evaporation and condensation are in equilibrium and occur at the same rate
- $P_w = P_{ws}$
- dewpoint = temperature
- wet bulb = dry bulb
- RH = 100%

Note: The only property that affects P_{ws} is temperature

P_{ws} Saturation Vapor Pressure

P_{ws} - maximum vapor pressure or amount of water vapor that can exist at a given temperature. Expressed in units of pressure.

$P_{ws} = 480 \text{ mbar}$



Relative Humidity (%)

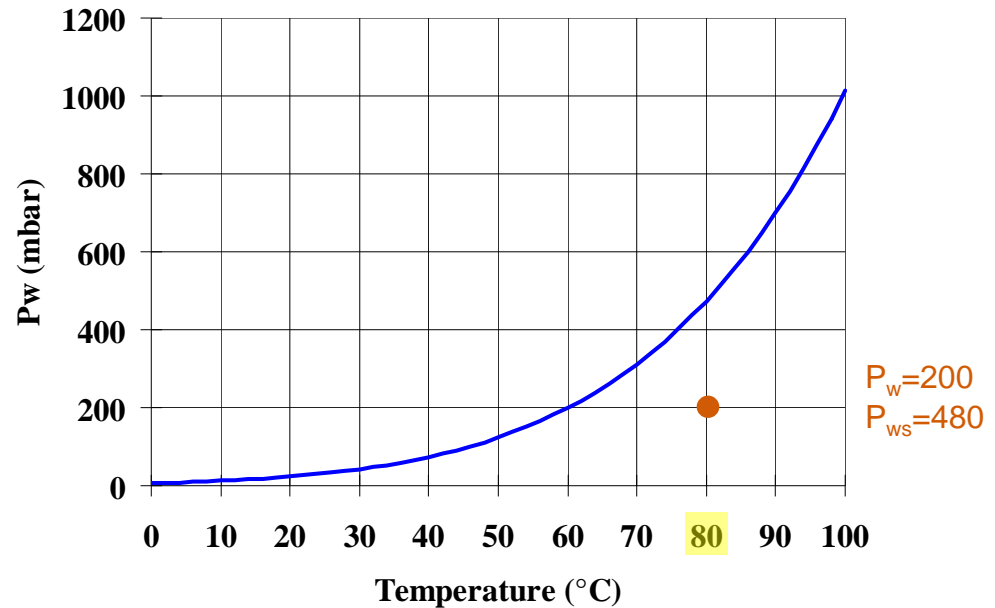
RH

Relative humidity is the ratio of water vapor partial pressure present in a gas (P_w) to the saturation vapor pressure of water at that temperature [$P_{ws}(t)$]

Relative humidity

$$\%RH = 100 \times \frac{P_w}{P_{ws}(t)}$$

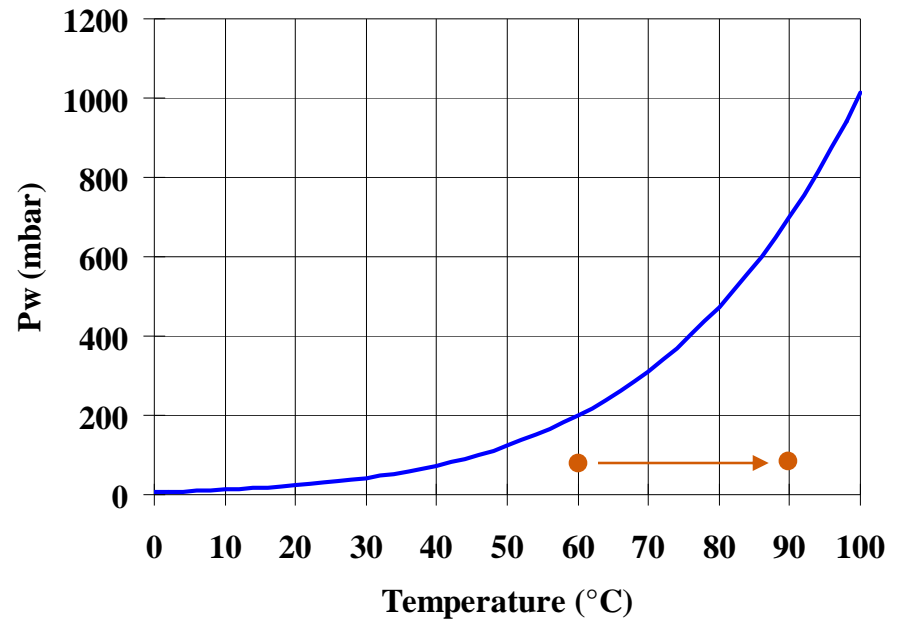
$$\%RH = 100 \times \frac{200}{480}_{(t=80)} = 42\%$$



Note: Relative humidity is strongly proportional to temperature and its measurement is very sensitive to temperature differences.

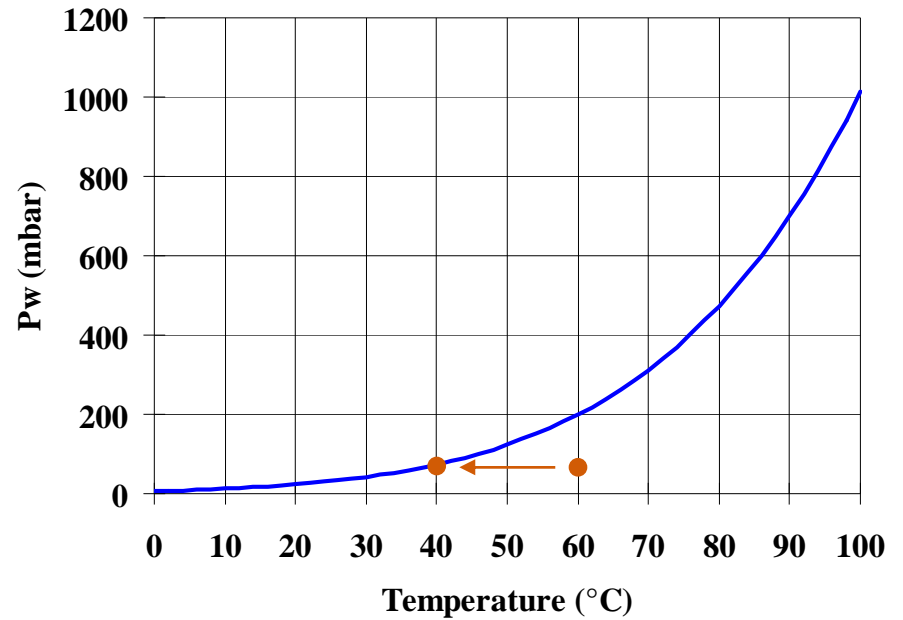
Temperature and Relative Humidity

$$\%RH = 100 \times P_w / P_{ws}(t)$$



Temperature and Relative Humidity

$$\%RH = 100 \times P_w / P_{ws}(t)$$



Temperature and Relative Humidity – Rule of Thumb #1

Rule of Thumb #1*

- As temperature increases, air becomes drier (RH decreases)
- As temperature decreases, air becomes wetter (RH increases)

- drier and wetter are relative terms; applies to a closed system where pressure and water vapor content do not change

What about pressure and Relative Humidity?

Recall Dalton's Law of Partial Pressures $P_t = P_w + P_{dry}$

If double total pressure;

then $2(P_t) = 2(P_w + P_{dry}) = 2P_w + 2P_{dry}$

so P_w changes proportionately to overall pressure changes

remember that P_w remains unchanged because T is unchanged

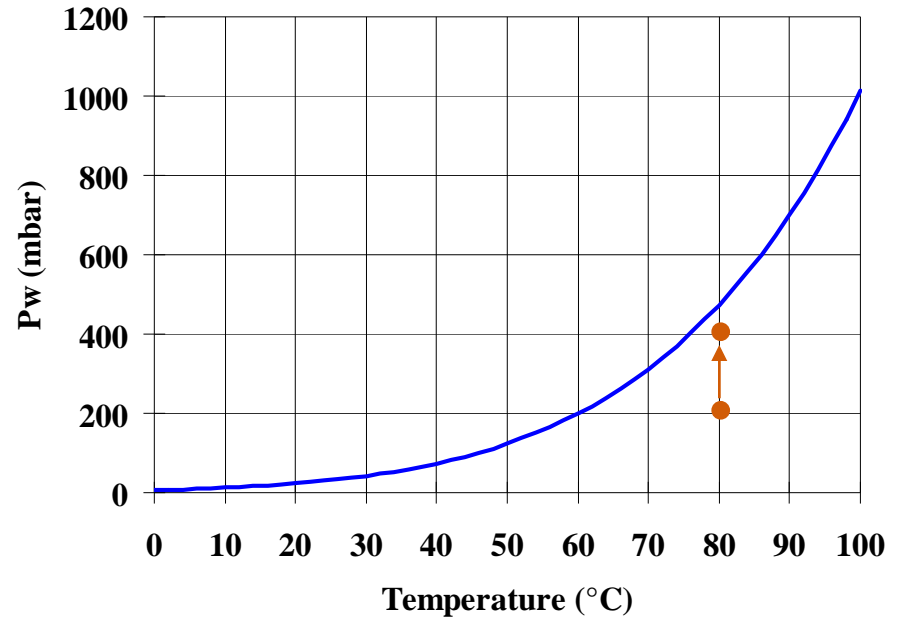
Pressure and Relative Humidity

$$P_t = 1000 \text{ mbar}$$

We double the total pressure

$$\text{so } P_t = 2000 \text{ mbar}$$

What happens to P_w ? P_{ws} ?



Pressure and Relative Humidity – Rule of Thumb #2

Rule of Thumb #2*

- As pressure decreases, air becomes drier (RH goes down)
- As pressure increases, air becomes wetter (RH goes up)

* drier and wetter are relative terms; applies to a closed system where temperature and water vapor content do not change

Relative Humidity Application Example



Relative humidity is the common parameter in HVAC/DCV applications where comfort balanced with efficiency is the main concern.

Summary & Questions

1. Water vapor theory
2. Dalton's law of partial pressures
3. P_w & P_{ws}
4. $RH = P_w/P_{ws}$
5. Temperature and RH – Rule of thumb
6. Pressure and RH – Rule of thumb

Vaisala GMP Humidity Resources

- On-line Humidity Calculator www.vaisala.com/humiditycalculator
- Slide Rule Calculator to order – <http://forms.vaisala.com/forms/RequestSlideRule>
- Psychrometric Chart - <http://forms.vaisala.com/forms/RequestPsychChart>
- Humidity Conversion Formulas - http://forms.vaisala.com/forms/humidity_conversion

For expert assistance with your humidity measurement

Email: instruments@vaisala.com

Direct telephone: 800-408-9454

Web: www.vaisala.com/instruments

Next Webinar – part 2 of 3

Thursday, April 14th, 10:00AM PDT

- Dewpoint & Frostpoint
- Humidity ratio or Mixing ratio
- ppm
- Absolute humidity
- Wet bulb temperature
- enthalpy

Everyone who registered for part 1 will get the invitation for part 2 automatically.

You will receive a follow up email with all of the resource links & link to recording.

Thank you!

This concludes the webinar.

Follow-up email will arrive shortly with the resource links & further contact information.

Link to the recorded version.